

### EXPLANATION

1000 Water table elevation (feet above sea level)  
 General direction of ground-water movement  
 Regional surface-water divide

### Aquifer

- Quaternary, unconfined
- Quaternary, confined
- Cretaceous
- Sioux Quartzite

### Data collection symbols

- Well measured for water level
- Well sampled for general chemistry
- Well sampled for tritium
- Well Sampled for <sup>14</sup>C, <sup>13</sup>C, <sup>3</sup>H, and <sup>18</sup>O; data are shown in Table 2.

Approximate depth to top of aquifer (feet) - not shown for wells measured only for water levels.

Water type

### Water types

- A - Calcium-magnesium-bicarbonate
- B - Calcium-magnesium-sulfate
- C - Sodium-sulfate

Surficial sand and gravel deposits - Map units sh, sd, sld, so, and sv as shown on Plate 1, Part A. Indicates approximate extent of surficial aquifers. Uncolored where water table occurs in materials that do not yield significant quantities of water.

### SURFICIAL HYDROGEOLOGY

By Randy Brady 1997

#### INTRODUCTION

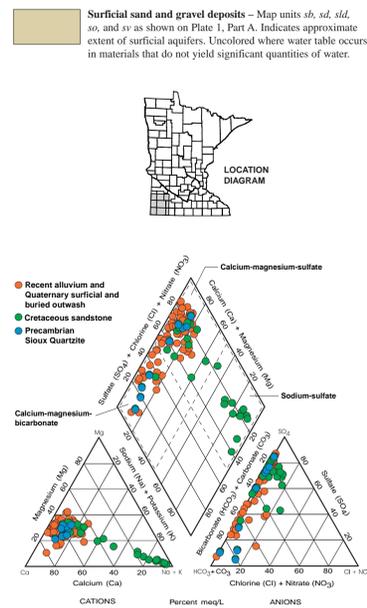
The Southwestern Minnesota Regional Hydrogeologic Assessment study area extends north one degree of latitude and east one degree of longitude from the southwestern corner of the state. The study area covers the 236 square mile area that includes all or part of nine counties, on a regional scale, ground-water occurrence, movement, and chemistry, with primary focus on ground water in surficial deposits. Data collected for this study include two synoptic water-level measurements of 230 domestic and 40 observation wells, and general water chemistry analyses for 107 wells. Samples were also collected for isotope analysis including tritium for 76 wells and <sup>14</sup>C and <sup>18</sup>O stable isotopes of hydrogen and oxygen for nine wells.

#### WATER TABLE

Construction of a water-table map usually relies on water levels obtained from wells located in a water-table aquifer. However, the water table in most of the study area exists in loam to clay loam till, materials generally not considered suitable as an aquifer. Therefore, water-table elevations are rarely found in these areas. However, the water table is commonly expressed at the surface in lakes, rivers, and wetlands. The water-table elevations for these features were obtained from U.S. Geological Survey (USGS) 1:24,000 topographic maps. Additional information included depth to water-table data from geophysical work by Department of Natural Resources staff in Rock and Pipestone counties. The available information suggests that the water table is everywhere within 50 feet of the land surface and approximates a subdued surface topography. Therefore, topographic contours from USGS 1:100,000 map sheets were used to guide contour placement in areas lacking water-table information. The water-table contours shown on the map provide a regional description of the water-table surface. The general direction of ground-water flow is at right angles to the contour lines toward lower elevations, as shown by the arrows on the map. However, the lack of control points and the scale at which the map was created preclude its use for determining depth to the water table.

Recharge to the water table occurs throughout the study area by infiltration of precipitation, surface runoff, and ground-water movement from adjacent areas. The ability of geologic materials to transmit water (hydraulic conductivity) is a major controlling factor of the recharge rate. Ground water in areas where recharge is rapid is more susceptible to contamination (see Plate 4).

The water-table map shows highest water-table elevations in the western part of the study area and along the regional surface water divide. The divide coincides with the Bemis Moraine along much of its length (see Plate 1, Part A). Regionally, ground water flows north and east of the divide toward discharge areas along the Minnesota River, and south and west of the divide toward the Missouri River. Locally, ground water discharges toward topographically low areas, wetlands, streams, and lakes.



### TABLE 2. Oxygen, hydrogen, and carbon isotope data for nine wells in the Southwestern Minnesota RHA study area.

(Samples collected during 1994 and 1995 by Minnesota Department of Natural Resources staff; per mil, parts per thousand; B.P., before present; T.U., tritium units; n.s., not sampled)

Unique Number	Well depth, feet	$\delta^{18}O$ , per mil	$\delta^2H$ , per mil	$\delta^{13}C$ , per mil	<sup>14</sup> C, % modern	Age, years	Tritium, TU
<b>Quaternary wells</b>							
158019	193	-10.42	-73.8	-15.3	0.141	13,000	n.s.
172117	180	-9.4	-63.2	-20.5	0.004	>35,000	<0.8
212954	390	-1.34	-7.9	-17.2	0.085	18,000	<0.8
500869	215	-9.61	-77.1	-12.9	0.300	5,000	n.s.
500885	320	-11.09	-70.6	-15.9	0.036	24,000	n.s.
<b>Cretaceous wells</b>							
193961	222	-11.33	-80.2	-17.9	0.012	34,000	n.s.
196755	350	-11.25	-80.8	-16.7	0.021	29,000	n.s.
403808	194	-11.61	-83.8	-17.3	0.003	>35,000	<0.8
481721	498	-10.70	-76.9	-20.1	0.043	100	<0.8

\* Age interpretation by Scott C. Alexander and Dr. E. Calvin Alexander, Jr., University of Minnesota, Department of Geology and Geophysics.  
 † Buried aquifer wells only.

### TABLE 1. Characteristics of natural waters by aquifer in the Southwestern Minnesota RHA study area.

(Samples collected during 1994 and 1995 by Minnesota Department of Natural Resources staff;  $\mu$ S/cm, microsiemens per centimeter; TDS, total dissolved solids; bicarbonate volatilization correction method; (Hem, 1992); °C, degrees centigrade; mV, millivolts; mg/L, milligrams per liter; TU, tritium units)

Aquifer	Conduc-tivity, $\mu$ S/cm	TDS, mg/L	Temp-erature, °C	pH	EH, mV	Dis-solved oxygen, mg/L	Alka-linity, mg/L	Ca, mg/L	Mg, mg/L	Na, mg/L	K, mg/L	Fe, mg/L	SO <sub>4</sub> , mg/L	Cl, mg/L	Nitrate, mg/L	Mn, mg/L	Sr, mg/L	B, mg/L	SiO <sub>2</sub> , mg/L	Tritium, TU
<b>Quaternary</b>																				
Mean	1515	1495	9.8	7.08	44	0.25	341	286	82.4	60.9	9.70	4.34	816	11.1	3.4	1.063	1.457	0.243	14.39	5.4
Median	1524	1543	9.7	7.11	49	<0.01	333	294	85.1	62.4	8.71	3.71	830	12.3	0.2	0.811	0.910	0.154	10.54	2.1
Minimum	426	338	8.4	5.95	-185	<0.01	227	69.7	17.4	7.94	30.32	3.71	0.01	0.021	0.239	0.052	0.162	2.99	0.8	
Maximum	2768	2731	14.4	7.69	432	9.11	502	592	168.0	185.2	128	16.77	1700	108	56.5	3.620	3.279	6.623	26.25	35.5
Standard deviation	549	618	0.9	0.25	116	1.17	63	121	31.5	43.9	14.5	3.62	456	19.9	10.5	0.893	0.691	0.146	2.56	9.0
Number of samples	75	72	75	75	75	75	75	72	72	72	72	72	72	72	72	72	72	19	72	57
<b>Cretaceous</b>																				
Mean	1854	1591	11.0	7.66	-13	0.06	295	158	54.8	277.8	9.35	1.95	861	40.8	0.3	0.464	1.821	1.116	8.36	<0.8
Median	1524	1543	9.7	7.11	49	<0.01	333	294	85.1	62.4	8.71	3.71	830	12.3	0.2	0.811	0.910	0.154	10.54	2.1
Minimum	853	571	9.4	6.98	-250	<0.01	174	7.94	32.4	30.32	3.71	0.01	0.021	0.239	0.052	0.162	2.99	0.8		
Maximum	3337	2749	14.1	8.54	309	0.88	425	456	167.8	141.0	16.1	10.91	1720	143	2.4	2.870	4.280	2.755	22.05	0.8
Standard deviation	581	554	1.3	0.51	116	0.17	69	134	47.8	209.9	3.58	4.08	471	0.6	0.811	1.127	1.106	5.75	0.0	
Number of samples	27	26	27	27	27	27	27	26	26	26	26	26	26	26	26	26	26	26	6	26
<b>Sioux Quartzite</b>																				
Mean	1164	1135	9.6	7.09	31	2.59	283	215	65.5	51.4	5.73	2.6	591	8.79	5.9	0.546	1.053	0.220	10.79	3.2
Median	1059	1094	9.6	7.10	20	0.08	290	216	62.0	53.0	2.7	534	6.01	0.910	0.154	0.54	1.054	0.24	10.54	2.1
Minimum	505	270	8.7	6.87	-148	<0.01	163	63.3	22.1	11.20	<0.707	0.01	0.049	0.39	<0.01	<0.003	0.229	<0.023	8.53	0.8
Maximum	2410	2516	10.7	7.33	195	9.52	364	399	155.0	122.7	10.0	7.0	1633	35.2	29.2	1.477	2.294	0.684	13.24	13.6
Standard deviation	598	612	0.7	0.13	112	3.50	61	130	45.5	4.06	2.4	591	11.2	10.6	0.533	0.722	0.254	1.55	4.5	
Number of samples	11	9	11	11	11	11	11	9	9	9	9	9	9	9	9	9	9	9	9	7

Digital data file modified from 1990 Census TIGER/Line Files of the U.S. Bureau of the Census (source scale 1:100,000). Digital hydrology data modified from 1989 Digital Line Graph (DLG) files from the U.S. Geological Survey (source scale 1:100,000). Digital base annotation by the Minnesota Geological Survey and the Minnesota Department of Natural Resources.

Universal Transverse Mercator projection, grid zone 15, 1983 North American Datum, map angle 1.987. Vertical datum is mean sea level. Compiled 1996.

GIS and cartography by Shawn Beyer, Malathi Bhattacharjee, Michael Schaber, Minnesota Department of Natural Resources, and Norman Anderson, Land Management Information Center, Minnesota Planning Office. Desktop publishing layout by Kim Anderson, Communications Media, Minnesota Department of Administration. Digital assembly by Nordic Press.

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#### AQUIFERS

Most wells in the study area are completed in Quaternary sand and gravel deposits. Locally important aquifers include Cretaceous sandstones and fractured Precambrian Sioux Quartzite. Precambrian rocks underlie the entire region (Figure 4, Plate 2, Part A). High-standing areas of Precambrian rocks are composed of Sioux Quartzite, which is at or near the surface in Pipestone and Rock counties. A subsurface ridge of Sioux Quartzite extends eastward from Pipestone County to northwestern Jackson County. The supply potential for the Sioux Quartzite aquifer is generally as little as a few to several tens of gallons per minute (gpm). The supply potential is quite variable depending on the hydraulic characteristics of fractures (Kanivetsky, 1978) and weathered zones in the quartzite, and the degree of hydraulic connection to the surface or to an overlying aquifer. These factors are difficult to identify or predict with existing information.

Cretaceous sediments overlie the Precambrian crystalline rocks (Setterholm, 1990) and are found in topographically low areas on the bedrock surface. These sedimentary deposits are composed primarily of interbedded shale, siltstone, and sandstone. In drill cuttings these deposits may be difficult to distinguish from glacial sediments. In the northeastern part of the study area, chiefly Lyon and Redwood counties, wells completed in Cretaceous deposits are common because the overlying Quaternary deposits are typically less than 100 feet thick and in many places lack a reliable confining unit. During each glacial retreat, water from the melting ice left behind sand and gravel deposits generally referred to as outwash. These deposits tend to form networks of low, narrow meltwater channels generally oriented parallel or perpendicular to northwest-southeast-trending end moraines. Subsequent glacial events buried these potential aquifers beneath confining materials including till, lacustrine sediments, loess (wind-blown deposits), or alluvial sediments. The last glacial retreat deposited surficial sand and gravel, mapped as stream sediment on Plate 1, Part A. These deposits are in many areas unconfined and form the beds of present-day streams. The extent of the mapped stream sediment approximates the extent of the surficial aquifer. Yields for Quaternary aquifers are extremely variable. High yields are generally associated with surficial sand and gravel deposits that can yield as much as 1,000 gpm. However, yields are generally less than 100 gpm (Adolphson, 1983; Kanivetsky, 1979).

#### WATER CHEMISTRY

Water samples for chemical analysis were collected from April 1994 to June 1995 from 72 wells completed in Quaternary deposits (only two samples represent unconfined sand and gravel deposits); 26 wells completed in Cretaceous sediments, and 9 wells completed in the Sioux Quartzite. The locations of the sampled wells, with the exception of 16 wells located just outside the study area, are indicated on the map. The results were used to characterize water from these aquifers, and to evaluate ground-water recharge processes. Table 1 summarizes the water chemistry by aquifer. Ground water contains dissolved minerals derived from the geologic materials (soil, till, etc.) through which it moves. Factors affecting ground-water chemistry include residence time, length of flow path, initial water chemistry, chemical reactions, and land use.

Generally, ground water in the study area is very hard with total dissolved solids (TDS) ranging from 270 milligrams per liter (mg/L) to 2,749 mg/L, with most samples exceeding 500 mg/L, the U.S. Environmental Protection Agency's (EPA) secondary (non-enforceable) standard for public drinking water supplies. Chemical constituents commonly exceeding EPA's secondary drinking water standard include sulfate, iron, and manganese. Excessive quantities of these chemicals may give water an objectionable taste or odor, stain laundry and porcelain, or even plug well screens.

EPA's primary (enforceable) public water supply standard for nitrate, 10 mg/L, was exceeded in less than 10 percent of samples analyzed. Seven samples from wells completed in Quaternary deposits and two samples from wells completed in Sioux Quartzite exceeded the standard. No samples from wells completed in Cretaceous sediments exceeded the EPA primary standard for nitrate.

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